



ACTUALIZACIONES / Review

ON THE SYNTHESIS OF VITAMIN D IN THE DARKNESS

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Abstract

Published evidence reports the existence of two routes for the transformation of 7-dehydrocholecalciferol into previtamin D_3 : a photochemical route with the participation of UVB photons and another route that occurs in the darkness. Several reports appear to support the presence of these two routes in some mammals, birds, nonvascular plants (e.g.: mosses), vascular plants (e.g.: angiosperms) and lichens. The reviewed evidence suggests that in the darkness, the synthesis of vitamin D_3 follows the same scheme of the photochemical pathway, but at a reduced rate respect to the synthesis under UVB radiation. The process of vitamin D synthesis in the dark, then, may be taken as an insurance for survival, at least for mammals and birds.

The low rate of the synthesis of vitamin D_3 in the absence of light produce low concentrations of vitamin D_3 metabolites in plasma. Long term survival under these circumstances might be possible through upregulation of vitamin D receptors (VDRs). In mole rats (South African rodents that live in the dark in underground tunnels), the reduced rate of vitamin D_3 synthesis produce low levels of

plasma vitamin D_3 and their metabolites 25(OH) D_3 and $1\alpha,25(OH)_2D_3$. The fact that K_d and β_{max} of the complex $1\alpha,25(OH)_2D_3$ -VDR from the intestinal mucosa, kidneys and the Harderian glands of the mole rat *Heterocephalus glaber* are significantly different in each one of these tissues, is interpreted as an indicator that the VDRs are, in each tissue, adapted to the maintenance of normal physiological functions.

Key words: vitamin D, darkness, receptors, regulation, synthesis in plants.

Resumen

SOBRE LA SÍNTESIS DE VITAMINA D EN LA OSCURIDAD

Varios trabajos publicados han informado que existen dos mecanismos para la transformación de 7-dehidrocoleciferol en previtamina D_3 : uno iniciado con el auxilio de fotones UVB y un segundo que ocurre en la oscuridad, sin el auxilio de radiación ultravioleta. Una serie de publicaciones contienen información que apoya la presencia de estos dos mecanismos en mamíferos, pájaros, plantas no vasculares (musgos), vasculares (angiospermas) y líquenes. La evidencia revisada sugiere que, en la oscuridad, la síntesis de vitamina D_3 sigue

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el mismo esquema que la ruta fotoquímica y generalmente ocurre a una tasa reducida respecto de la síntesis bajo irradiación ultravioleta. La operación de la síntesis de vitamina D₃ en la oscuridad, por lo menos para mamíferos y pájaros, puede tomarse como un reaseguro de sobrevivida.

La reducida tasa de síntesis de vitamina D en ausencia de luz produce bajas concentraciones en plasma de los metabolitos de la vitamina. La sobrevivida saludable en estas condiciones sería posible mediante upregulation de los receptores. En las ratas topo *Heterocephalus glaber* (roedores sudafricanos que viven en

permanente oscuridad, en túneles subterráneos), la reducida tasa de síntesis de vitamina D₃ es la causa de los bajos niveles plasmáticos de la vitamina y sus metabolitos: 25(OH)D₃ y 1α,25OH₂D₃. El hecho de que el K_d y β_{max} del complejo 1α,25(OH)₂D₃-Vitamina-D-Receptor del intestino, riñón y glándulas de Harder de *Heterocephalus glaber* sean significativamente diferentes entre sí indicaría que los receptores se han modificado para mantener funciones fisiológicas normales en cada tejido.

Palabras clave: vitamina D, oscuridad, regulación, receptores, síntesis en vegetales

Introduction

During the industrial revolution, started in Europe in the 18th century, an outbreak of Bone Deforming Disease (rickets) appeared in children. The possible relationship between the industrialization of Northern Europe and rickets was reported by Sniadecki in 1822.¹ He observed that children living in rural areas did not suffer from rickets while children born and raised in Warsaw were plagued with the disease, and attributed the high incidence of rickets, in Polish children to their lack of sun exposure. In 1919, Huldschinski² reported the observation that children with rickets had a dramatic healing of their disease after several months of exposure to radiation from a mercury arc lamp. Two years later Hess and Unger³ reported that exposure to sunlight was an effective treatment for rickets. In 1935, 7-dehydrocholesterol was first isolated by Windaus *et al.* and vitamin D₃ was identified two years later by Windaus and Bock.⁵ The fact that irradiation of 7-dehydrocholesterol produced vitamin D₃ was finally proven in 1978 by Esvelt *et al.*⁶ through isolation and identification of vitamin D₃ by mass spectrometry. Details of the complete history of the cure of rick-

ets by irradiation to the discovery of vitamin D₃ and the effects of its metabolites can be read in a review by De Luca.⁷

For many years, only animal products were considered as sources of vitamin D₃. Today we know that vitamin D₃ and their metabolites 25(OH)D₃ and 1α,25OH₂D₃ are synthesized as well by nonvascular plants e.g.: mosses, vascular ones e.g.: angiosperms and lichens (a symbiosis between a fungus and an alga or cyanobacteria).⁸ Planktonic microalgae⁹ are small microscopic photosynthetic aquatic plants that produce vitamin D and are the source of vitamin D for fish. Microalgae are important for life on earth. They contain the photosynthetic pigments chlorophyll a and c; they produce approximately half of the atmospheric oxygen and simultaneously use carbon dioxide to grow photo-autotrophically.

The physiological function of vitamin D₃ in plants is yet poorly known. Plants have calcium channels and pumps similar to those found in animals¹⁰ and calcium ions are a core regulator of plant cell physiology.¹¹ Calcium is required for stimulation of growth, root initiation and promotion of germination in plants.¹² 1α,25(OH)₂D₃ has been shown to in-



fluence growth and calcium transport in roots of *Phaseolus vulgaris* (common bean)¹⁴ by increasing synthesis of calmodulin.¹³ a calcium-binding messenger protein found in all eukaryotic cells. The presence of tissue vitamin D₃ receptors (VDR) has been reported in *Solanum glaucophyllum* (waxyleaf nightshade).¹⁴

Synthesis of vitamin D in the skin

In the skin, the synthesis of vitamin D₃ has two steps (Figure 1): It starts with Δ^7 -dehydrocholesterol which is converted first into pre-vitamin D₃, which later isomerizes to vitamin D₃. This two-reaction process was thoroughly investigated by Holick and coworkers.¹⁵

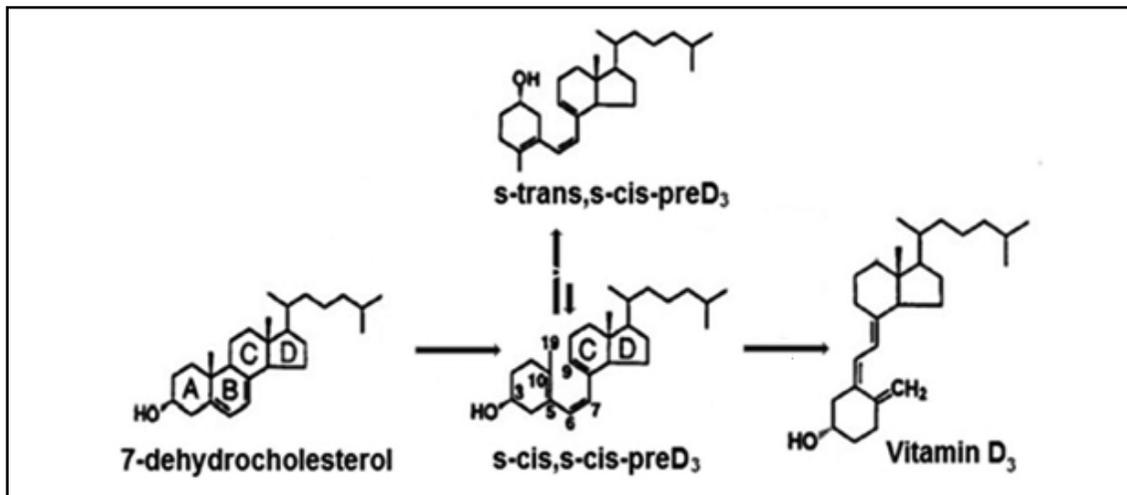


Figure 1. Photolysis of Δ^7 -dehydrocholesterol into s-cis,s-cis- previtamin D₃ and its thermal isomerization to vitamin D₃. Adapted from Wacker & Holick.¹⁶

First reaction: UVB photons excite the 7-dehydrocholesterol molecule and transforms it into a secosteroid (a steroid with a “broken” ring): the pre-vitamin D₃, a molecule that adopt two configurations: s-cis,s-cis- previtamin D₃ and s-trans,s-cis pre vitamin D₃.

Wacker and Holick¹⁶ postulated that in an organic solvent, such as hexane, previtamin D₃ preferentially exists in the s-trans,s-cis conformation, which is thermodynamically more stable. The less stable conformation: s-cis,s-cis- previtamin D₃ in hexane at 25°C isomerizes into vitamin D₃ with a half-life time of 91 h.

Second reaction: in the skin, previtamin D₃ is maintained in the s-cis, s-cis conformation, by interaction with membrane fatty acids. The

s-cis,s-cis- pre vitamin D₃ isomerize rapidly into vitamin D₃, with a T_{1/2} = 8 h. Much of the information about the previtamin D₃ isomerization to vitamin D₃ was obtained from experiments using organic solvents (e.g., hexane) to assess the conversion^{16,17} and assumed to reproduce the same phenomena occurring in human skin. The reaction rate of this isomerization is a direct function of temperature. Earlier studies found that this process was not affected by acids, bases, catalysts, or inhibitors of radical chain processes. Holick et al.¹⁷ concluded that the isomerization was a rearrangement within the molecule of previtamin D₃. There is no evidence for the existence of an enzymatic process in the skin that can

convert previtamin D₃ to vitamin D₃. During the synthesis of vitamin D₃, the hydrophilic and hydrophobic interactions of the s-cis, s-cis-previtamin D₃ with the membrane fatty acids are disrupted, thereby facilitating the ejection of vitamin D₃ from the skin cell membrane into the extracellular space. It enters the circulation by binding to the serum vitamin D₃ binding protein DBP. The removal of vitamin D₃ from the skin as it is being produced changes the isomerization reaction from a reversible to an irreversible process.¹⁵⁻¹⁷ This would explain the relatively rapid rise in the serum vitamin concentration after UVB exposure.

The importance of membrane microenvironments on the isomerization previtamin D₃ \rightleftharpoons vitamin D₃ received further support from a kinetic study in an aqueous solution of β -cyclodextrin.¹⁸ Cyclodextrins, a group of naturally occurring, truncated cone-shaped oligosaccharides, have a unique ability to complex a variety of foreign molecules into their hydrophobic cavities in aqueous solution including vitamin D₃. With this model, Tian and Holick¹⁸ demonstrated that, at 5°C, the forward and reversed rate constants for the isomerization of previtamin D₃ \rightleftharpoons vitamin D₃ were increased by more than 40 and 600 times, respectively, compared with those in n-hexane.

Transference of vitamin D₃ from the skin to the circulation. After previtamin D₃ is thermally isomerized to vitamin D₃, the latter is transported to the dermal capillary bed beneath the dermo-epidermal junction. The mechanism of translocation is largely unknown. In an attempt to understand this event Tian et al.¹⁹ studied the kinetics of vitamin D₃ formation and the time course of appearance of vitamin in the circulation after exposure of chickens to UVB radiation. Their data indicate a fast rate of formation of vitamin D₃ from previtamin D₃ and a relatively fast rate of translocation from skin to circulation. By examining the time course of appearance of vitamin D₃ in circulation, they found a rapid

phase of vitamin D₃ appearance from 8 to about 30 h post-irradiation and a relatively slower phase of its disappearance after the peak of vitamin D₃ in the circulation. No previtamin D₃ could be detected 1 h after UVB irradiation. Thus, only vitamin D₃ is preferentially removed from skin into circulation.

The importance of DBP in the translocation of vitamin D₃ into circulation was defined by Haddad et al.²⁰ They investigated the transport of skin-synthesized vitamin D₃ into circulation in seven healthy volunteers who received whole-body irradiation by assessing the time course distribution of plasma protein-bound vitamin D₃ in high (>1.3 g/ml) and low density (<1.3 g/ml) fractions after UVB irradiation. They found that plasma vitamin D₃ concentration began to increase 10 h after irradiation, peaked at 24 h, and lasted for a week in the high-density layer where all the DBP was present. These observations indicated that the endogenously photosynthesized vitamin D₃ circulates in serum bound to the DBP. The latter phenomenon differs from the orally administered vitamin D₂²⁰, which is evenly distributed between the high- and low-density layers at 4, 8, and 24 h after the ingestion.

Evidence on the synthesis of vitamin D₃ in the darkness

Researchers have evaluated the synthesis of vitamin D in the darkness measuring vitamin D₃ and/or its hydroxylated metabolites as the end point of their experiments.

A. Mammals

Damara mole rats (*Fukomys damarensis*) live in underground tunnels (Pitcher et al.²¹). The plasma levels of 25(OH)D₃ and 1 α ,25(OH)₂D₃ of these animals increase when exposed to sunlight or supplemented with vitamin D₃ (Table 1). These experiments demonstrated that Damara mole rats have the capacity to produce vitamin D₃ and its metabolites without the participation of sunlight.



Table 1. Plasma levels (mean \pm SEM) of 25(OH)D₃ and 1 α ,25(OH)₂D₃ in controls, vitamin D₃ supplemented and exposed to sunlight Damara mole rats.

Metabolites	Controls no sunlight exposure n=10	Vit D ₃ supplemented, n=5 2.5 ng/g of food eaten	Sunlight exposure n=5 15 min/day/3 weeks
25(OH)D ₃ , ng/ml	< 5	30 \pm 3	< 5
1 α ,25(OH) ₂ D ₃ , pg/ml	12 \pm 2	35 \pm 2	25 \pm 3

Table 2. Saturation analysis of 1 α ,25(OH)₂D₃ binding to VDRs from the intestinal mucosa, kidney, Harderian glands and skin of naked mole rats. (Sergeev et al., 1993²²)

	Tissue	β_{max} (fmoles/mg protein)	K _d (pM)
Control mole rats	Intestine mucosa	380 \pm 51.7	748 \pm 148
	Kidney	79.9 \pm 9.4 ^b	183 \pm 20.2 ^b
	Harderian glands	28.1 \pm 4.2 ^{b,c}	97.0 \pm 12.2 ^{b,c}
	Skin	<1.5	ND
Mole rats supplemented with a high dose of vitamin D ₃ 10 days before assay	Intestine mucosa	599 \pm 90.7 ^a	813 \pm 93.9
	Kidney	158 \pm 27.6 ^{a,b}	206 \pm 9.0 ^b
	Harderian glands	32.3 \pm 5.0 ^{b,c}	81.7 \pm 7.6 ^{b,c}
	Skin	<1.5	ND

ND: not determined. Results are means \pm SEM. P<0.05: ^a: from the control group, ^b: from the intestinal mucosa in the same group, ^c: from the kidney in the same group.

Table 2 has been transcribed from the paper of Sergeev et al.²² to point out that low plasma levels of 1 α ,25(OH)₂D₃ may be compensated by upregulation of the VDRs. Table 2 displays K_d values (inversely related to of the drug affinity for the binding site) and the maximum number of binding sites per mg of receptor protein (β_{max}), of naked mole rats (*Heterocephalus glaber*). These investigators succeeded in upregulating VDR, administering a high dose of vitamin D₃. This acute treatment increased the β_{max} from the intestinal mucosa and kidney tissue, did not affect the K_d values of VDRs of the intestinal mucosa, kidney and Harderian

glands and increased the 24-hydroxylase activity in the kidneys (vitamin D₃ treated mole rats: 64.0 \pm 09.9 pmol/h/g tissue; vitamin D₃ deficient animals: 16.3 \pm 1.37 pmol/h/g tissue; P<0.05).

The fur of fur-bearing animals had been thought to be comparable to clothing in humans, which prevents vitamin D₃ synthesis in the skin during exposure to sunlight. The work of Hymøller and Jensen,²³ demonstrated otherwise. Sixteen Danish Holstein dairy cows were deprived of vitamin D₃ for 6 months by omitting the vitamin D₃ in their feed and housing them without access to sunlight. They were subjected to 4 degrees of coverage of their

bodies with a fabric that prevented vitamin D₃ synthesis in the covered skin areas. The experimental groups were: a) cows without any coverage (100% exposed body surface area), b) cows fitted with udder covers (94% exposed body surface area), c) cows fitted with horse blankets (28% exposed body surface area) and d) cows fitted with both horse blankets and udder covers (24% exposed body surface area). The cows were let out to pasture daily between 10.00 and 15.00 h for 4 weeks. Blood samples were collected during the study and analyzed for their 25-hydroxyvitamin D₃ content.

The results (Figure 2) showed that uncovered cows had a higher 25(OH)D₃ concentration in plasma after 28 days of access to sunlight compared with covered cows and that the estimated peak plasma concentration of 25(OH)D₃ at the end of the experiment was inversely correlated to the body surface area covered ($R^2 = 0.99556$, $P < 0.0022$). It was concluded that cows, like humans, synthesize vitamin D₃ evenly over their body surface.

B. Birds

The feathers of feather-bearing birds has been thought to be comparable to clothing in humans, which prevents vitamin D₃ synthesis in the skin during exposure to sunlight. Two hypotheses predominate among those proposed for the origin of feathers: they evolved for flight or for thermal insulation (McGowan, 1989²⁴).

Kale et al.²⁵ (Tables 3 and 4) has reported the vitamin D status of a flightless nocturnal bird, the brown kiwi (*Apteryx mantelli*). Kiwis caught in the wild had significantly low plasma levels of 25(OH)D₃ (3.0nmol/L, n=4). These birds feed mainly on earthworms (Reid et al.²⁶) that contain undetectable concentrations of Vitamin D₃ (Finke²⁷), with small amounts of plant matter (Reid et al.²⁶). No significant differences were observed in plasma 25(OH)D₃ concentrations between those kept captive in outdoor enclosures receiving natural sunlight (19.8±10.9, n=6) and those kept in nocturnal houses receiving no supplemental UV-B light (19.7±9.5, n=6). Captive birds are fed on a mixture of in-

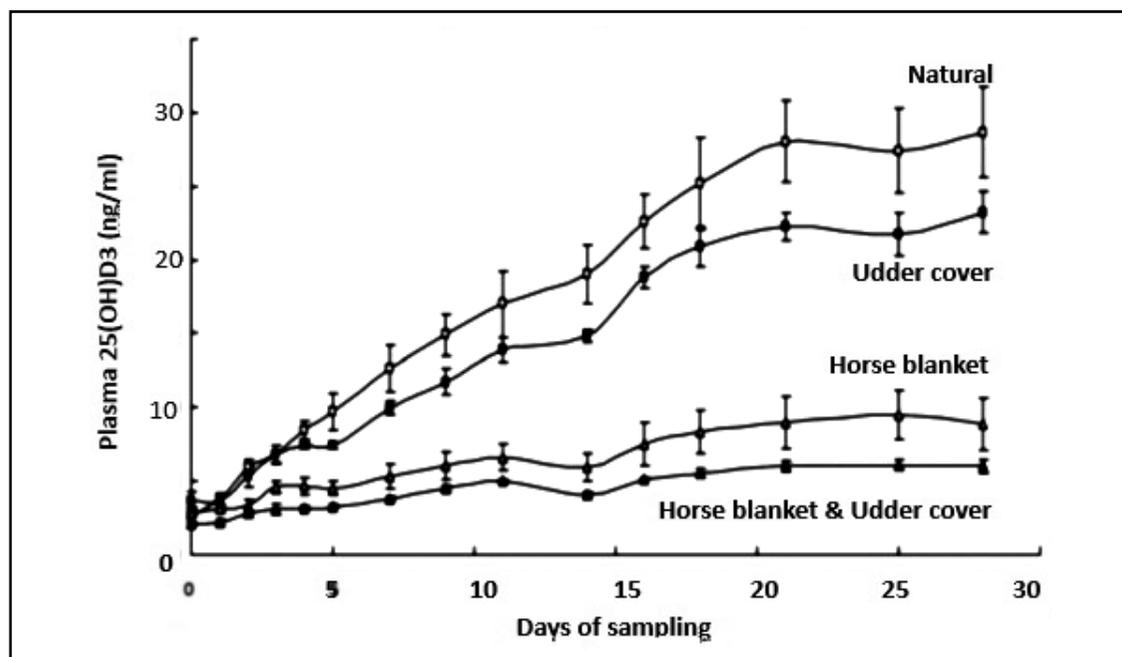


Figure 2. Plasma concentration of 25(OH) vitamin D₃ of dairy cows, after 6 months of vitamin D₃ deprivation and exposed to sunlight without (natural) and three degrees of body surface covering (see text for further details). Redrawn from Hymøller and Jensen.²³



Table 3. Plasma 25(OH)D₃, ng/ml (mean ± SEM) of kiwi, tuatara and New Zealand sea lions (Kale et al. 2018).²⁵

Species	Plasma 25(OH)D ₃
Kiwi, wild (<i>Apteryx mantelli</i>)	3.0
Tuatara, captive (<i>Sphenodon punctatus</i>)	104.2±12.8
New Zealand sea lions (<i>Phocarctos hookeri</i>)	104.4±10.0

Table 4. Concentration of vitamin D₃ (µg of vit D₃/g of fresh skin tissue) after excision of tissue samples from individuals of three species. Kale et al. 2018.²⁵

Species	n	not irradiated	after 8 hours UVB*
Kiwi	4	<0.03	0.038
Tuatara	4	<0.03	0.084
Sea Lion	2	0.50	1.60

*5 J/cm²

sects, vegetables, fruit, minced ox heart, and nutritional supplements, including a premix that contains vitamin D (Minson²⁸), which explains their plasma levels of 25(OH)D₃.

Kale et al.²⁵ also performed *in vitro* experiments with the skins of kiwis to gain information on this site of synthesis of vitamin D₃. Skin samples were prepared in a dark room. Any fat or feathers present were removed. For each bird, two pieces of skin, two 10 cm² pieces of skin were dissected. One piece was exposed to UVB radiation for 8 h, equivalent to 5 J/cm² of accumulated UV light, in a reflector box at 37 °C. The skin was moistened with phosphate-buffered saline at 1 h intervals. The control skin sample was kept wrapped in aluminum foil at 4 °C. Each sample was then cut into 5 × 5 mm pieces in a dark room, placed in a light proof bag, weighed and freeze-dried. Freeze-dried samples were extracted and analyzed for Vitamin D₃ contents. Analysis of extracts were done by reverse phase high performance liquid chromatography. Non-irradiated kiwi skins

had Vitamin D₃ levels below the detection limit of the assay: <0.03 µg of vit D₃/g of fresh skin tissue. There was, however, a measurable increase in response to UV exposure, indicating the existence of the dermal synthesis pathway: 0.038 µg of vit D₃/g of fresh skin tissue.

Kiwi skins produced small but measurable amounts of Vitamin D₃ after UV irradiation. Analysis of the characteristics of their VDRs might confirm an adaptation to physiological low levels of vitamin D₃ similar to that of mole rats.

C. Plants

S. glaucophyllum is a plant native of South America. The ingestion of its leaves causes calcinosis in cattle (Worker and Carrillo 29). Early studies correlated this property to the existence in the plant of vitamin D₃, 15(OH)₂D₃ and 1α,25(OH)₂D₃ (Wasserman et al.,³⁰ Espareza et al.³¹). Curino et al.³² carried out experiments with *S. glaucophyllum* tissue (callus). The term “callus” refers to an undifferentiated

plant cell mass grown on a culture medium, to produce genetically identical cells. For calli cultures, leaf explants from plants collected in the field, thoroughly disinfected, were inoculated in agar-solidified Murashige-Skoog medium supplemented with 2,4-dichlorophenoxyacetic acid and kinetin and grown at 25°C under complete darkness. Calli began to develop after 15 days of culture. After two months of subculture all the surface of culture flasks was covered with tissue. Calli grown for at least 4 months in the dark were placed in liquid nitrogen and submitted to lipids extraction. Figure 3 displays the results of Curino et al.³², the $1\alpha,25(\text{HO})_2\text{D}_3$ contents of *S. glaucophyllum* calli grown in the darkness vs. calli grown in the darkness + timed UV irradiation. The most interesting result of these experiments is the significant amount of $1\alpha,25(\text{HO})_2\text{D}_3$ in calli grown in the darkness and not the relatively small increase of the metabolite under UV irradiation.

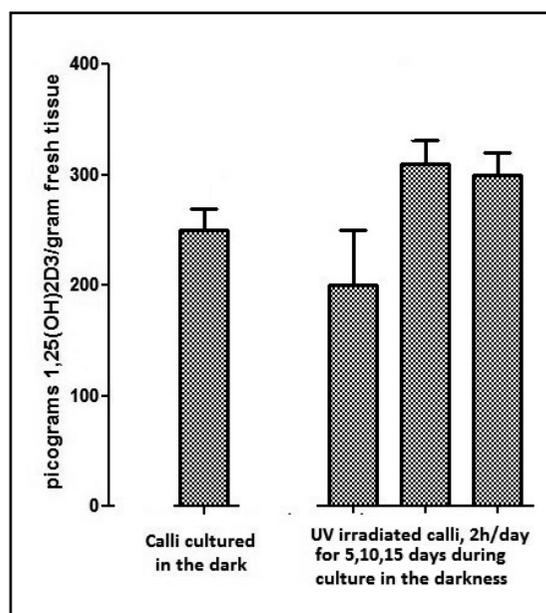


Figure 3. Tissue content of $1\alpha,25(\text{HO})_2\text{D}_3$ of *S. glaucophyllum* cells cultured in the darkness and in the darkness+UV irradiation. Redrawn from Curino et al.³²

Björn and Wang³³ confirmed a UVB dependent synthesis of vitamins D₂ and D₃ in the leaves of the tomato plant. Plants were grown in a greenhouse with or without UVB radiation 0.85 kJ plant weighted UVB radiation/m².per day (Table 5).

Table 5. Contents of provitamins and vitamins D₂ and D₃ µg/g of dry matter in tomato plants *Solanum lycopersicum* (Björn and Wang, 2001).³³

	Not irradiated	Irradiated
Provitamin D ₂	1.83	2.23
Vitamin D ₂	0	0.087
Provitamin D ₃	0.61	0.76
Vitamina D ₃	0	0.28

An interesting observation is that the provitamin D₃ content of tomato leaves is not reduced by growing plants under UV-B radiation, although a substantial amount of vitamin D₃ is formed. Björn and Wang³³ assume the existence of a feedback mechanism regulating the amount of the provitamins.

The evidence reported by Jäpelt et al.^{34,35} (Table 6) agrees with the above mentioned results. Note that the absence of irradiation does not cancel the synthesis of vitamin D₃ and its metabolites and that UVB irradiation increases the content of vitamin D₃ and its metabolites.

As shown by Wang et al.³⁶ provitamin D₂, vitamin D₂ and vitamin D₃ have been identified in the thallus of the lichen species *Cladina arbuscula*.

Inspection of Figure 4 and Table 7, indicate that the contents of vitamins D₂ and D₃ in *Cladina arbuscula* decrease as a function of the latitude. There is a significant positive correlation between the vitamins D contents



Table 6. Contents of 7-dehydrocholesterol, vitamin D₃, 25(OH)D₃ and 1,25(OH)₂D₃ (ng/g of dry matter) of plants cultured with a 16/8 light/dark cycle, with or without additional irradiation with an UVB source (30 min per day for 7 days). Data from Jäpelt et al.^{34, 35}

Species	7-Dehydrocholesterol	Vitamin D ₃	25(OH)D ₃	1,25(OH) ₂ D ₃
Controls				
<i>Solanum lycopersicum</i>	470	1.7	<0.02	<0.1
<i>Solanum glaucophyllum</i>	670	3.2	0.8	<0.1
<i>Capsicum annuum</i>	30	<0.02	<0.02	<0.1
UVB-irradiated				
<i>Solanum lycopersicum</i>	230	100	4.3	<0.1
<i>Solanum glaucophyllum</i>	1260	200	31	32
<i>Capsicum annuum</i>	30	2.9	0.5	<0.1

Table 7. Ranges of provitamin D₂, vitamin D₂ and vitamin D₃ contents (µg/g of dry weight of *Cladina arbuscula*) measured by high performance liquid chromatography, electrospray ionization mass spectrometry and UV spectrophotometry Wang et. al. 2001.³⁶⁾

Place	Approx. latitude	provitamin D ₂	Vitamin D ₂	Vitamin D ₃
Finland	67°N	91 - 99	0.23 - 0.49	0.67 - 2.04
Sweden	56°N	106 - 120	0.43 - 0.54	1.04 - 1.24
Greece	40°N	149	0.57	1.73 - 1.84

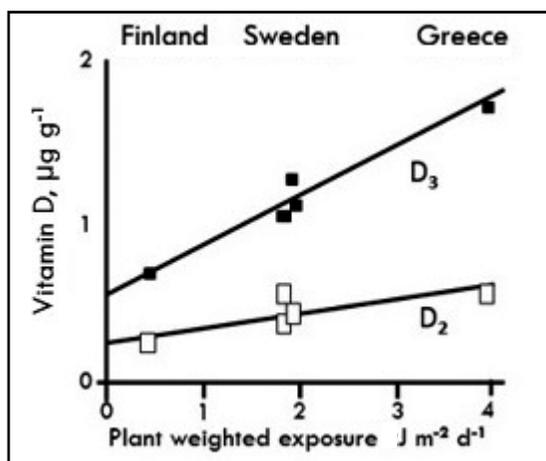


Figure 4. Left: linear relationship between the vitamins D₂ and vitamin D₃ contents of *C. arbuscula* as a function of the weighted UV exposure. Latitude of sampling places: Finland: 66-67°N, Sweden: 56°N and Greece: 40°N. Redrawn from Wang et al.³⁶⁾

and the estimated radiant energy received by lichens collected at three different latitudes. Björn et al.³³ believe that the lichen might function as receptors of UV radiation. In agreement with the observations of Jäpelt et al.,^{34,35} the contents of vitamins increase with the amount of radiant energy. Inspection of the Figure 4 indicate that vitamins content of *Cladina* are not zero at zero irradiation, which is interpreted to support the hypothesis that the synthesis of vitamin D₃ may proceed both with and without UV irradiation.

The coexistence of the two vitamins D in *Cladina* (the contents of vitamin D₃ is greater than that of D₂) may be explained by the fact that lichens are composite organisms that arise from algae or cyanobacteria living with fungi in a symbiotic relationship.

Comments on the survival of terrestrial species that live in the dark

The reviewed evidence on the synthesis of vitamin D₃ in the darkness, allows the following conclusions:

a) It uses the same components of the photochemical pathway (Figure 1).

b) It has a reduced rate respect to the synthesis under UVB irradiation.

The identification of the mechanism/s by which certain species, though lacking access to sunlight, produce enough vitamin D₃ to meet their physiological functions is an as yet unsolved matter. Assessed by actual human standards, mole rats and fruit bats^{37,38} are vitamin D deficient. The natural habitats of both species exclude sunlight and their diet is herbivorous with no obvious source of vitamin D₃.

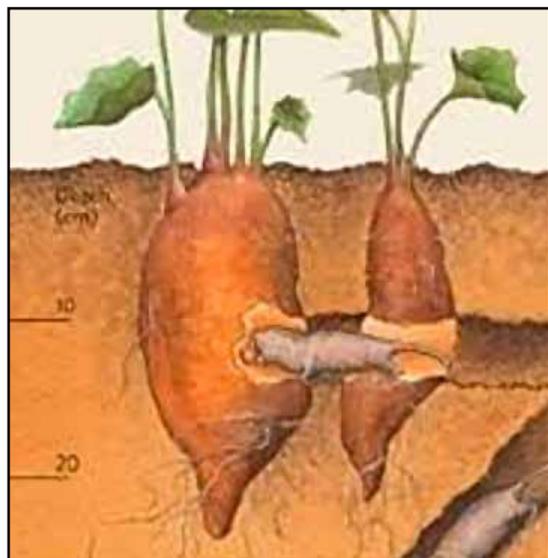


Figure 5. Scheme of the hábitat of mole rats (upper-left corner of a figure in the paper by Park and Buffenstein³⁹).

Figure 5 suggests that mole rats feed on the roots of plants. This reviewer could not find a survey of the plants with subterranean enlarged organs (enlarged to store energy in the form of carbohydrates) in the

areas of mole rats colonies. The amount of 1 α ,25(OH)₂D₃ that can be obtained from this source is probably negligible but it is as yet unknown.

With the commercial availability of [³H]1 α ,25(OH)₂D₃ as a radioligand of high specific activity, the presence of a specific binding site for 1 α ,25(OH)₂D₃ in low numbers (~ 100 fmol/mg protein) have been reported in *Solanum glaucophyllum* leaves, berries, stems and roots (Weissenberg et al).⁴⁰ The protein binding of [³H]-vitamin D₃ has been demonstrated in *Phaseolus vulgaris* (French beans) roots.⁴¹

The low rate of the synthesis of vitamin D₃ in the absence of light produce low concentrations of vitamin D₃ metabolites in plasma. Long term survival under these circumstances might be possible through upregulation of VDRs. The fact that the K_d of the complex 1 α ,25(OH)₂D₃-VDR from the intestinal mucosa, kidneys and the Harderian glands of *Heterocephalus glaber* (south African mole rats that live in the dark in networks of underground tunnels) are significantly different for each one these tissues (Table 2),²² is interpreted to indicate that the VDRs are, in each tissue, adapted to the maintenance of normal physiological functions.

Described in organic chemistry terms, the photochemical production of vitamin D₃ (Figure 1) is a conrotatory^A electrocyclic^B opening of the B-ring of 7-dehydrocholesterol to provide the triene^C previtamin D₃, which then undergoes a thermally-induced sigmatropic^D rearrangement to yield vitamin D₃.⁴² It is of interest that *sigmatropic reactions usually do not require a catalyst*. A steric interaction between the C ring and the carbon 19 methyl group of s-cis, s-cis-previtamin D₃, could produce a conformer of previtamin D₃ energetically unfavorable and therefore less stable. As a result, a rotation around the carbon 5 and carbon 6 single bond occurs and an energetically more stable s-trans, s-cis-previtamin D₃ is formed (Figure 1). The unstable s-cis, s-cis-previtamin D₃ configuration can undergo an intramolecular hydrogen rearrangement to form vitamin D₃.



Which is/are the element/s that, in the darkness, would replace the energy of UVB photons for the opening of ring B? Hypothetically, they might be replaced by a) some as yet undetermined fraction of the electromagnetic component of cosmic rays or b) by the existence of a specific binding protein for 7-dehydrocholesterol. Experimental proof is needed to support these hypotheses.

Cosmic rays have three kinds of components: muons and hadrons (particles that have rest masses) and photons of the electromagnetic spectrum: radio waves, microwaves, infrared, visible light, ultraviolet, X-rays, and gamma rays. The ultraviolet radiation has

a wavelength range 10 to 400 nm.^{43,44} It appears possible that penetrating UV radiation, attenuated by the interaction with leather or feathers reach the cells containing 7-dehydrocholesterol and may contribute to the synthesis of vitamin D₃. Inspection of the action spectrum for the conversion of 7-dehydrocholesterol to previtamin D₃ (Figure 6), reveals that the optimum wavelengths for the production of previtamin D₃ lie between 295 and 300 nm with a maximum at 297 nm. It also indicate that wavelengths below 295 nm (though with lower efficiency) can break the C9-C10 bond of 7-dehydrocholesterol and produce previtamin D₃ (Figure 6).⁴⁵

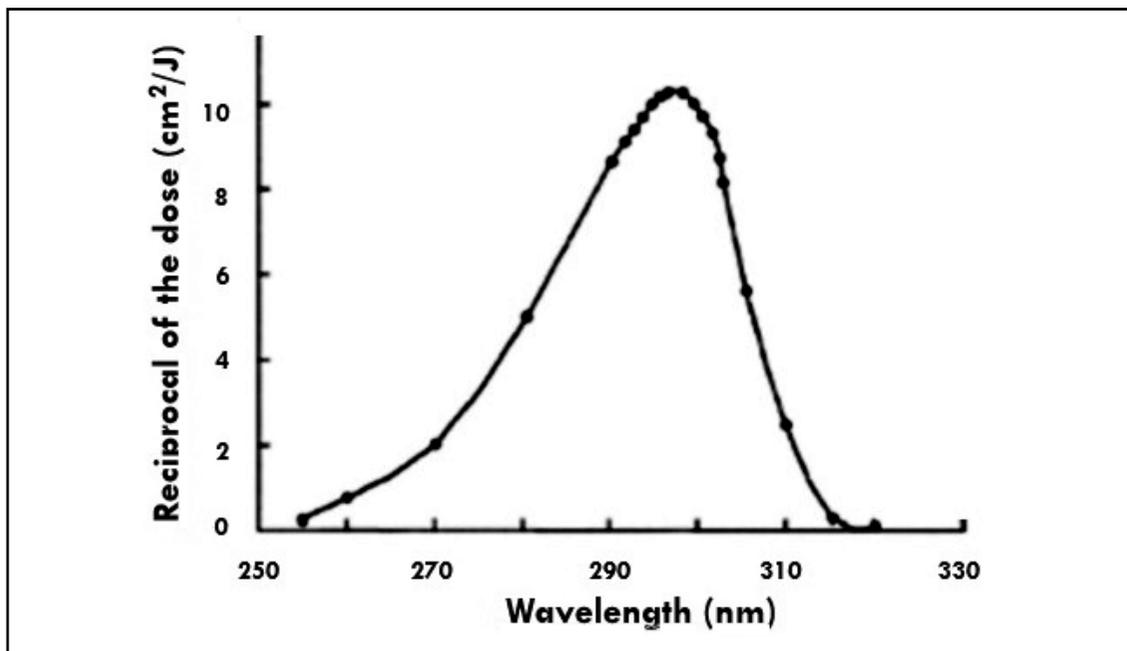


Figure 6. Action spectrum for the conversion of 7-dehydrocholesterol to vitamin D₃ in the human skin. Redrawn from MacLaughlin et al.⁴⁵

Other possible but as yet not experimentally demonstrated mechanisms for the synthesis of vitamin D₃ in the darkness

Norman and Norman,⁴⁶ centering their attention on the opening of the B-ring of 7-dehydrocholesterol have stated: “A major

unresolved problem concerns identification of the biological mechanisms by which certain animals possessing an endoskeleton but lacking access to sunlight, produce sufficient amounts of vitamin D₃ to meet their physiological needs. This includes many

sea fish, which live in the ocean below the level of sunlight penetration (10-20 m) and ...have massive quantities... of vitamin D₃ in their livers". They report that the chorismate mutase (an enzyme present in microorganisms) acting on chorismic acid produced a retro-ene-rearrangement similar to that transforming 7-dehydrocholesterol into provitamin D₃. In simple words, an enzyme may replace the lack of light.

They also stated: "Steroid biosynthesis have been shown to proceed from squalene via a selective epoxidation and acid catalyzed olefinic cyclization. It therefore seem reasonable to suggests that such pathways have also evolved in animals lacking access to light, to catalyze the conversion of 7-dehydrocholesterol to vitamin D₃". They describe three mechanisms based on the existence enzymes demonstrated in the synthesis of steroids in vertebrates. They conclude that "It should be appreciated that the four mechanisms offered here are suggestions based upon an understanding of organic chemistry; there is no evidence to support their existence".⁴⁶

Footnotes

A Conrotatory: An electrocyclic reaction can either be classified as conrotatory or disrotatory based on the rotation at each end of the molecule. In conrotatory mode, both

atomic orbitals of the end groups turn in the same direction such as both atomic orbitals rotating clockwise or counter-clockwise.

B Pericyclic: a reaction in which bonds are made or broken in a concerted cyclic transition state. A concerted reaction is one which involves no intermediates during the course of the reaction

C triene: an unsaturated hydrocarbon containing three double bonds between carbon atoms

D sigmatropic: the name *sigmatropic* is the result of a compounding of the sigma designation from single carbon-carbon bonds and the Greek word *tropos*, meaning turn. A sigmatropic reaction in organic chemistry is a pericyclic reaction wherein the net result is one σ -bond is changed to another σ -bond in an uncatalyzed the reaction intra-molecular process. In this type of rearrangement reaction, a substituent moves from one part of a π -bonded system [alternating single and double bonds] to another part in an intra-molecular reaction with simultaneous rearrangement of the π system.

Conflicto de intereses: el autor declara no tener conflicto de intereses.

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